

**(DME 215)**

**B. Tech. DEGREE EXAMINATION, MAY - 2015**

**(Examination at the end of Second Year)**

**MECHANICAL ENGINEERING**

**Paper - V : Basic Thermodynamics**

**Time : 3 Hours**

**Maximum Marks : 75**

**Answer question No. 1 compulsory**

**(15)**

**Answer ONE question from each unit**

**(4 x 15 = 60)**

- 1) a) Write about property and state of a substance.
- b) Define work and heat transfer.
- c) What is mean by exergy and dead state?
- d) Define classius statements.
- e) State Carnot cycle.
- f) What is PMM1 and PMM2?
- g) Define critical point temperature. When it exists?

**Unit - I**

- 2) a) What do you mean by macroscopic and microscopic view points? Differentiate each other.
- b) Write the differences between system and control volume.

OR

- 3) A gas expands from an initial state where the pressure is 340KPA and the volume is 0.0425 m<sup>3</sup> to a final pressure of 136 KPA. The relationship between pressure and volume of the gas  $pv^2$  is constant. Determine the work for the process.

**Unit - II**

- 4) a) State the limitations of first law of thermodynamics.
- b) Define enthalpy? Why does the enthalpy of an ideal gas?

- c) An engine operating on a Carnot cycle works with in temperature limits of 600K and 300K. If the engine receives 2000KJ of heat, evaluate the work done and thermal efficiency of the engine.

OR

- 5) a) What is steady flow energy equation for nozzle and throttling device explain briefly?  
b) Define specific heats at constant volume and constant pressure and hence deduce a relation between two.

### Unit - III

- 6) a) Give Kelvin Planck statement and clausius statements of the second law.  
b) Calculate available energy in 40kg of water at 75°C with respect to the surroundings at 5°C the pressure of water begin 1atm.

OR

- 7) Air of mass 0.5 kg as an ideal gas executes a carnot cycle having a thermal efficiency of 50%. The heat transfer to the air during the isothermal expansion 40 KJ. At the beginning of isothermal expansion the pressure is 7 bar and the volume is 0.12 m<sup>3</sup>. Determine :  
a) The maximum and minimum temperature for the cycle.  
b) The volume at the end of isothermal expansion.  
c) The work and heat transfer for each of four processes.

### Unit - IV

- 8) a) An engine working on the otto cycle is supplied with air at 0.1MPa, 35°C. The compression ratio is 8. Heat supplied is 2100 KJ/KG. Calculate the maximum pressure and temperature of the cycle and the mean effective pressure.  
(for air  $C_p = 1.005$ ,  $C_v = 0.718$  and  $R = 0.287$  KJ/KgK.  
b) A diesel engine has a compression ratio of 14 and cut off takes place at 6% of the stroke. Find the air standard efficiency.

OR

- 9) a) What is mean by pure substance? Explain T-S diagram for a pure substance.  
b) Steam at 10 bar and 3000C passing through a convergent-divergent nozzle expands reversibly and adiabatic ally till the pressure falls to 2 bar. If the velocity of the steam entering into the nozzle is 50m/sec. Determine the exit velocity of the steam.

